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Key indicators

Single-crystal X-ray study T = 298 K Mean σ (C–C) = 0.004 Å R factor = 0.050 wR factor = 0.095 Data-to-parameter ratio = 8.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-Amino-4-(4-methoxyphenyl)-6-(piperidin-1-yl)pyridine-3,5-dicarbonitrile

The crystal structure of the title compound, $C_{19}H_{19}N_5O$, is stabilized by intermolecular $C-H\cdots O$ hydrogen bonds and $C-H\cdots \pi$ interactions.

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Comment

The structure of the title compound, (I) (Fig. 1), is similar to that of 2-amino-4-(4-methoxyphenyl)-6-(pyrrolidin-1-yl)pyridine-3,5-dicarbonitrile (Raghukumar *et al.*, 2003). The methoxyphenyl ring is twisted with respect to the pyridine ring, with a dihedral angle of 43.20 (1)° between them. The piperidine ring adopts a half-chair conformation. The crystal packing is consolidated by intermolecular $C-H\cdots O$ hydrogen bonds and $C-H\cdots \pi$ interactions (Table 1 and Fig. 2).



Experimental

Piperidine (0.38 g, 4.4 mmol) was added dropwise to a solution of a 4methoxybenzaldehyde (0.50 g, 3.67 mmol) and malononitrile (0.50 g,



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Figure 1

The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.

7.35 mmol) in 5 ml of anhydrous methanol at 273 K. The reaction mixture was then stirred at room temperature for 10 h. The resulting mixture was purified by chromatography on silica using petroleum ether/acetone ($5:1\nu/\nu$) as eluant to obtain the title compound as a gray-yellow solid (yield 0.39 g, 33%). The product was recrystallized from ethanol at room temperature to give crystals suitable for single-crystal X-ray diffraction.

Z = 2

 $D_x = 1.330 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

 $\mu = 0.09 \text{ mm}^{-1}$

T = 298 (2) K

 $R_{\rm int} = 0.144$

 $\theta_{\rm max} = 27.5^{\circ}$

Block colorless

 $0.18 \times 0.10 \times 0.10 \; \mathrm{mm}$

9714 measured reflections

2051 independent reflections

1618 reflections with $I > 2\sigma(I)$

Crystal data

 $\begin{array}{l} C_{19}H_{19}N_5O\\ M_r = 333.39\\ Monoclinic, P2_1\\ a = 9.4430 \ (7) \ A\\ b = 7.5675 \ (6) \ Å\\ c = 12.5714 \ (9) \ Å\\ \beta = 112.06^\circ\\ V = 832.58 \ (11) \ Å^3 \end{array}$

Data collection

Bruker SMART CCD area-detector
diffractometer
φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 2001)
$T_{\min} = 0.985, T_{\max} = 0.991$

Refinement

Refinement on F^2	H atoms treated by a mixture of
$R[F^2 > 2\sigma(F^2)] = 0.050$	independent and constrained
$wR(F^2) = 0.095$	refinement
S = 0.94	$w = 1/[\sigma^2(F_0^2) + (0.0449P)^2]$
2051 reflections	where $P = (F_0^2 + 2F_c^2)/3$
233 parameters	$(\Delta/\sigma)_{\rm max} < 0.001$
	$\Delta \rho_{\rm max} = 0.28 \text{ e} \text{ Å}^{-3}$
	$\Delta \rho_{\rm min} = -0.39 \text{ e} \text{ Å}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$\overline{\begin{array}{c} C1 - H1B \cdots O1^{i} \\ C19 - H19C \cdots Cg3^{ii} \end{array}}$	0.97	2.51	3.228 (3)	131
	0.96	2.72	3.345 (4)	123

Symmetry codes: (i) x - 1, y, z - 1; (ii) -x + 1, $y - \frac{1}{2}$, -z + 1. Cg3 is the centroid of the benzene ring

We tried to obtain crystals suitable for single-crystal X-ray diffraction from three different solvents. Unfortunately, the single crystals were all of poor quality, hence the high value of R_{int} . In the absence of significant anomalous dispersion effects, 1733 Friedel pairs were merged. H atoms bound to N5 were located in a difference



Figure 2

Part of the packing of (I), with hydrogen bonds and $C-H\cdots\pi$ interactions drawn as dashed lines. For clarity, H atoms not involved in these interactions have been omitted.

Fourier map and refined with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm N})$ and the restraint N-H = 0.86 (1) Å. Other H atoms were positioned geometrically and treated as riding, with C-H = 0.93 Å, $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}$ (C) for aromatic, C-H = 0.97 Å, $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$ for CH₂, and C-H = 0.96 Å, $U_{\rm iso}({\rm H}) = 1.5U_{\rm eq}({\rm C})$ for CH₃ atoms.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT-Plus* (Bruker, 2001); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON*.

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